



## Section 01 - Product And Company Identification

Product Identifier Sodium Sulphite, Technical Grade

Sodium Sulphite, Catalyzed

Other Means of Identification

Product Use and Restrictions on Use Bleach operations, reducing agent in the manufacture of dyes, photographic

developers and fixers, food additive, water treatment dechlorinating agent.

Initial Supplier Identifier ClearTech Industries Inc.

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Prepared By ClearTech Industries Inc. Technical Department

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## Section 02 - Hazard Identification

#### GHS-Classification

Acute Toxicity-Oral Category 4
Acute Toxicity-Inhalation Category 5
Skin Corrosion/Irritation Category 2
Serious Eye Damage/Eye Irritation Category 2A
Respiratory and Skin Sensitization Category 1
STOT-Single Exposure Category 3
Aspiration Hazard Category 2



### **Signal Word**

Warning

#### **Hazard Statements**

Harmful if swallowed.

May be harmful if inhaled.

Causes skin irritation.

Causes eye irritation.

May cause allergy or asthma symptoms or breathing difficulties if inhaled.

May cause an allergic skin reaction.

May cause respiratory irritation.

May be harmful if swallowed and enters airways.

### **Physical Hazards**

No known physical hazards.

### **Pictograms**





### **Precautionary Statements**

Do not eat, drink or smoke when using this product.

Wash thoroughly after handling.

Wear protective gloves and eye/face protection as specified by the manufacturer/supplier or the competent authority.

Avoid breathing dust/fume/ gas/mist/vapours/spray.

Contaminated work clothing should not be allowed out of the workplace.

In case of inadequate ventilation wear respiratory protection.

Use only outdoors or in well-ventilated area.

IF SWALLOWED: Immediately call a POISON CENTER or doctor/physician.

Do NOT induce vomiting.

IF INHALED: If breathing is difficult, remove to fresh air and keep at rest in a position comfortable for breathing.

If experiencing respiratory symptoms call a POISON CENTER or doctor/physician.

IF ON SKIN: Wash with plenty of soap and water.

Take off contaminated clothing and wash before re-use.

If skin irritation or rash occurs, seek medical advice/attention.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

If eye irritation persists, get medical advice/attention.

Wash hands after handling.

Store locked up.

Store container tightly closed in well-ventilated place.

## **Section 03 - Composition / Information on Ingredients**

Chemical Name	CAS Number	Weight %	Unique Identifiers
Technical:	7757 92 7	079/	
Sodium Sulphite	7757-83-7	97%	



Sodium Sulphate 7757-82-6 <3%

Catalyzed:

 Sodium Sulphite
 7757-83-7
 >95%

 Sodium Sulphate
 7757-82-6
 <2%</td>

 Cobalt Sulphate
 10124-43-3
 0.04%

Common Name and Synonyms Disodium sulfite, sodium sulfite

**Section 04 - First Aid Measures** 

**Inhalation** Remove victim to fresh air. Give artificial respiration only if breathing has stopped.

If breathing is difficult, give oxygen. Seek immediate medical attention.

Skin Contact / Absorption Remove contaminated clothing. Wash affected area with soap and water. Seek

medical attention if irritation occurs or persists.

Eye Contact Contact lenses should never be worn when working with this product. Flush

immediately with water for at least 20 minutes. Forcibly hold eyelids apart to ensure

complete irrigation of eye tissue. Seek immediate medical attention.

**Ingestion** If conscious, immediately give 2 to 4 glasses or water or milk and induce vomiting

under medical supervision. Get immediate medical attention.

Additional Information Treat symptomatically. Note potential for anaphylactic shock with allergic

individuals.

**Section 05 - Fire Fighting Measures** 

Suitable Extinguishing Media Product does not burn. Where fire is involved, use any fire fighting agent

appropriate for surrounding material; use water spray to cool fire-exposed

surfaces.

Unsuitable Extinguishing Media Not Available

**Specific Hazards During Fire Fighting** During a fire, irritating/toxic sulphur oxides may be generated.

Special Protective Equipment for Fire- Wear NIOSH-approved self-contained breathing apparatus and protective clothing.

**Fighters** 

Further Information Not Available



### **Section 06 - Accidental Release Measures**

Personal **Precautions** / **Protective** Wear appropriate personal protective equipment. Ventilate area. Only enter area **Equipment / Emergency Procedures** 

with PPE. Stop or reduce leak if safe to do so.

**Environmental Precautions** Prevent material from entering sewers.

Methods and Materials for Containment Pick up and arrange disposal without creating dust. Sweep up and shovel. Do not flush with water. Keep in suitable, closed containers for disposal. and Cleaning Up

## Section 07 - Handling and Storage

**Precautions for Safe Handling** Use proper equipment for lifting and transporting all containers. Use sensible

industrial hygiene and housekeeping practices. Wash thoroughly after handling.

Avoid all situations that could lead to harmful exposure.

Store in a cool, dry area and away from acids or oxidizers. Keep container closed. **Conditions for Safe Storage** 

Protect from physical damage.

Incompatibilities Acids and strong oxidizers.

## **Section 08 - Exposure Controls and Personal Protection**

osure Limit(s)			
Component	Regulation	Type of Listing	Value
Sodium sulphite	Not Established		
Sulphur dioxide	OSHA	TWA	5ppm
	ACGIH	TLV	5ppm
	ACGIH	STEL	2ppm
Cobalt sulphate	OSHA	PEL	0.1mg/m <sup>3</sup>
	ACGIH	TLV	0.2mg/m <sup>3</sup>
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### **Engineering Control(s)**

Mechanical ventilation (dilution or local exhaust), process or personnel enclosure **Ventilation Requirements** and control of process conditions must be provided in accordance with all fire codes

and regulatory requirements. Supply sufficient replacement air to make up for air

removed by exhaust systems.



Other Emergency shower and eyewash must be available and tested in accordance with

regulations and be in close proximity.

**Protective Equipment** 

**Eyes/Face** Chemical goggles, full-face shield, or a full-face respirator is to be worn at all times

when product is handled. Contact lenses should not be worn; they may contribute to

severe eye injury.

Hand Protection Impervious gloves of chemically resistant material (rubber or PVC) should be worn

at all times. Wash contaminated clothing and dry thoroughly before reuse.

Skin and Body Protection Body suite, aprons, and/or coveralls of chemical resistant material should be worn at

all times. Wash contaminated clothing and dry thoroughly before reuse.

Impervious boots of chemically resistant material should be worn at all times. No

special footwear is required other than what is mandated at place of work.

Respiratory Protection If dusty conditions prevail use a dust mask approved by NIOSH. If sulphur dioxide

should be released use a self-contained breathing apparatus.

Thermal Hazards Not Available

Section 09 - Physical and Chemical Properties

**Appearance** 

Physical State Solid

**Colour** White to pale yellow crystals or powder

**Odour** Odourless

Odour Threshold Not Applicable

**Property** 

pH 9.8 for a 5% solution

Melting Point/Freezing Point Decomposes

Initial Boiling Point and Boiling Range Not Applicable

Flash Point Not Applicable



Evaporation Rate Not Applicable

Flammability Non-Flammable

Upper Flammable Limit Not Applicable

Lower Flammable Limit Not Applicable

Vapour Pressure (mm Hg, 20°C) Not Applicable

Vapour Density (Air=1) Not Applicable

Relative Density Not Available

Solubility(ies) 17% w/w at 10℃, 28% w/w at 33℃ in water

Partition Coefficient: n-octanol/water Not Available

Auto-ignition Temperature Not Applicable

**Decomposition Temperature** 900℃

Viscosity Not Available

Explosive Properties Non-Explosive

Specific Gravity (Water=1) 2.63

% Volatiles by Volume Not Applicable

Formula Na<sub>2</sub>SO<sub>3</sub>

Molecular Weight 126.04

## Section 10 - Stability and Reactivity

Reactivity Reaction with strong oxidizers cause vigorous exothermic reactions. Reaction with

acids release sulfur dioxide gas. Sodium sulphide residue forms when product

decomposes.

**Stability** Stable under normal conditions.



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**Possibility of Hazardous Reactions** Hazardous polymerization will not occur. Toxic gas is formed with acid contact.

**Conditions to Avoid** Air contact, heat, moisture, incompatible materials.

Oral I Dea

**Incompatible Materials** Acids and strong oxidizers.

Hazardous Decomposition Products Sulphur dioxide gas, sodium sulphide.

## **Section 11 - Toxicological Information**

## **Acute Toxicity**

Component

Component	Oral LD <sub>50</sub>	Dermai LD <sub>50</sub>	LC <sub>50</sub>	
Sodium Sulphite	820mg/kg (mouse)	Not Available	5500mg/m³ (rat, inhalation, 4hr)	
	3560mg/kg (rat)			
Sodium Sulphate	5989mg/kg (mouse)	Not Available	Not Available	
Cobalt Sulphate	424mg/kg (rat)	Not Available	Not Available	
Chronic Toxicity – Carcino	genicity			
Comp	Component IARC		IARC	
Sodium	sulphite	Group 3: Not classifiable as to its carcinogenicity to		
Sodium sulphate		Group 3: Not classifiable as to its carcinogenicity to humans.		
Cobalt	Cobalt sulphate Group 2B: Possibly carcinoger		y carcinogenic to humans.	
Skin Corrosion/Irritation	Dust may cause	Dust may cause skin irritation from prolonged contact. Solutions will irritate		
Serious Eye Damage/Irritation	rious Eye Damage/Irritation Dust may irritate or burn eyes. Solutions will irritate or burn eyes.		ate or burn eyes.	
Ingestion	some asthmatics	Ingestion may irritate gastrointestinal tract. May cause severe allergic reaction some asthmatics and sulphite sensitive individuals. Large doses may cause viole colic and diarrhea, circulatory disturbances, central nervous system depression even death.		
Inhalation	Inhalation of dust	Inhalation of dust may irritate respiratory tract.		
Respiratory or Skin Sensitiza	tion May cause allerg	May cause allergic skin and respiratory sensitization reactions.		
Germ Cell Mutagenicity		Sodium sulfite has been demonstrated to be mutagenic in microbial systems; nowever, it is not mutagenic in studies involving insects and is not considered to		

Dermal I Dra



present a mutagenic threat to multicellular organisms.

Reproductive Toxicity Not Available

STOT-Single Exposure Inhalation of dust may irritate respiratory tract.

STOT-Repeated Exposure May cause damage to the following organs: peripheral nervous system, central

nervous system (CNS).

Aspiration Hazard Prolonged or repeated inhalation may cause chronic irritation, inflammation,

delayed pulmonary edema, and alteration of sense of smell and taste.

Synergistic Materials Not Available

## **Section 12 - Ecological Information**

Ecotoxicity			
Component	Toxicity to Algae	Toxicity to Fish	Toxicity to Daphnia and Other Aquatic Invertebrates
Sodium Sulphite	Not Available	LC <sub>50</sub> (Western Mosquito fish, 96 hour): 660mg/L	LC <sub>50</sub> (Daphnia magna, 48 hour): 440ppm
Sodium Sulphate	Not Available	LC <sub>50</sub> (Fathead minnow, 96 hour): 13500-14000mg/L	LC <sub>50</sub> (Daphnia magna, 48 hour): 2564ppm
Biodegradability	Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise. The products of degradation are less toxic than the product itself.		
Bioaccumulation	Bioaccumulation	is not expected to occur.	
Mobility	Not Available		
Other Adverse Effects	Not Available		

## **Section 13 - Disposal Considerations**

Waste From Residues/Unused Products

Dispose in accordance with all federal, provincial, and/or local regulations including the Canadian Environmental Protection Act.

**Contaminated Packaging** 

Dispose in accordance with all federal, provincial, and/or local regulations including the Canadian Environmental Protection Act.



## **Section 14 - Transport Information**

UN Number Not Regulated

UN Proper Shipping Name Not Regulated

Transport Hazard Class(es) Not Regulated

Packaging Group Not Regulated

**Environmental Hazards** 

**Special Precautions** 

**Transport in Bulk** 

**TDG** 

Other Secure containers (full and/or empty) with suitable hold down devises during

shipment and ensure all caps, valves, or closures are secured in the closed

position.

PRODUCT CLASSIFICATION: This product has been classified on the preparation date specified at section 16 of this MSDS / SDS, for transportation in accordance with the requirements of part 2 of the Transportation of Dangerous Goods Regulations. If applicable, testing and/or published test data regarding the classification of this product are listed in the references at section 16 of this MSDS / SDS.

## **Section 15 - Regulatory Information**

NOTE: THE PRODUCT LISTED ON THIS SDS HAS BEEN CLASSIFIED IN ACCORDANCE WITH THE HAZARD CRITERIA OF THE CANADIAN CONTROLLED PRODUCTS REGULATIONS. THIS SDS CONTAINS ALL INFORMATION REQUIRED BY THOSE REGULATIONS.

### **Section 16 - Other Information**

### **Preparation Date** July 21, 2014

**Note:** The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations.

### Attention: Receiver of the chemical goods / SDS coordinator

As part of our commitment to the Canadian Association of Chemical Distributors (CACD) Responsible Distribution<sup>®</sup> initiative, ClearTech Industries Inc. and its associated companies require, as a condition of sale, that you forward the attached Safety Data



Sheet(s) to all affected employees, customers, and end-users. ClearTech will send any available supplementary handling, health, and safety information to you at your request.

If you have any questions or concerns please call our customer service center or technical service department.

#### **Abbreviations**

ACGIH American Conference of Governmental Industrial Hygienists

AIHA American Industrial Hygiene Association

CAS Chemical Abstract Service
DSL Domestic Substance List
EC Effective Concentration

IARC International Agency for Research on Cancer IDHL International Digest of Health Legislation

LC Lethal Concentration LD Lethal Dosage

MSHA Mine Safety and Health Administration

NIOSH National Institute for Occupational Safety and Health

NFPA National Fire Protection Association NTP National Toxicology Program (U.S.A.)

OSHA Occupational Safety and Health Administration (U.S.A.)

PEL Permissible Exposure Limit
PPE Personal Protection Equipment
STEL Short-term Exposure Limit

STOT Specific Target Organ Systemic Toxicity

TLV Threshold Limit Value

TSCA Toxic Substances Control Act TWA Time Weighted Average

UN United Nations

WEEL Workplace Environmental Exposure Level

WHMIS Workplace Hazardous Materials Information System

#### References

- 1) CHEMINFO: Sodium sulfate. (2014). Retrieved from Canadian Centre for Occupational Health and Safety: http://ccinfoweb2.ccohs.ca/cheminfo/records/4028E.html
- 2) CHEMINFO: Sodium sulfite. (2014). Retrieved from Canadian Centre for Occupational Health and Safety: http://ccinfoweb2.ccohs.ca/cheminfo/records/3422E.html
- 3) Sigma-Aldrich. (2012, April 27). Material Safety Data Sheet: Sodium sulfite. Oakville, ON. Retrieved from Sigma-Aldrich.
- 4) U.S. National Library of Medicine. (2002). *Sodium sulfite*. Retrieved from TOXNET: http://toxnet.nlm.nih.gov/cgi-bin/sis/search2/f?./temp/~VzDcCy:1

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# 24 Hour Emergency Number - All Locations - 1(306) 664-2522 Alternative - 1(800) 387-7503

**End of Safety Data Sheet** 

